POSSIBLE PARTICIPATION OF MEMBRANE THIOL GROUPS ON THE MECHANISM OF NAD(P)+-STIMULATED Ca2+ EFFLUX FROM MITOCHONDRIA

Anibal E. Vercesi

Departamento de Bioquímica, IB, Universidade Estadual de Campinas, 13100 Campinas, São Paulo, Brazil

Received January 11, 1984

SUMMARY: NAD(P) $^+$ -stimulated Ca $^{2+}$ efflux from mitochondria is inhibited by bongkrekate and slightly stimulated by carboxyatractylate. Addition of oxaloacetate, an NAD(P) oxidant, or diamide, a thiol oxidant, to de-energized mitochondria incubated in Ca $^{2+}$ -free medium induced a small decrease in turbidity of the mitochondrial suspension compatible with small structural changes of mitochondria. Similar to NADP+-stimulated Ca $^{2+}$ efflux these changes were also inhibited by bongkrekate and slightly stimulated by carboxyatractylate. The similarity between the effects of oxaloacetate and diamide, on both Ca $^{2+}$ efflux and mitochondrial structure, indicates the existence of a common denominator, possibly the oxidation of specific thiol groups, regarding the mechanism by which these agents stimulate Ca $^{2+}$ efflux from mitochondria.

We have demonstrated that ruthenium red-insensitive Ca²⁺ efflux from isolated liver (1) and heart (2) mitochondria is stimulated by the oxidized steady-state of matrix NAD(P)H. Although these results were confirmed by other laboratories with isolated mitochondria (3-10), intact cells (11,12) and isolated liver (13), there is no agreement regarding the biological relevance (5,6,10) and the molecular mechanism of such a process (7,14,15).

Regarding the factors controlling the rate of Ca²⁺ efflux from isolated mitochondria some reports have indicated that small structural changes of mitochondria (from aggregated to

Abreviations: EGTA, ethylene glycol bis (β -aminoethyl ether) N,N'-tetraacetic acid; HEPES, 4-(2-hydroxyethyl)-1-piperazine ethane-sulfonic acid; TMPD, N,N,N',N'-tetramethyl-p-phenylene diamine; OAA, oxaloacetate; BOH, β -hydroxybutyrate; BA, bong-krekate; CAT, carboxyatractyloside; DIAMIDE, diazenedicarboxylic acid bis-dimethylamide.

orthodox configuration) induced by Ca²⁺ and other agents increase the membrane permeability to the cation (16-18). Bongkrekate and atractylate, both inhibitors of the adenine nucleotide translocase, induce opposite effects on mitochondrial configuration (19). Bongkrekate causes a change of the mitochondria to a more condensed configuration and enhances their ability to retain Ca²⁺ under certain conditions (20-23). Atractylate, on the other hand, stimulated Ca2+release from mitochondria (16,21,23). Other data indicate that the redox state of membrane thiol groups also affects the membrane permeability to Ca2+ (9,22,24). In fact, diamide, an oxidant of membrane-bound thiol groups, stimulates Ca²⁺ efflux and dithioerythritol a thiol reductant decreases the rate of Ca^{2+} efflux from mitochondria (24). The results shown in this communication indicate that NADP +- stimulated Ca2+ efflux from mitochondria is correlated to both a Ca²⁺-independent change in mitochondrial configuration as evidenced by a decrease in turbidity of the mitochondrial suspension, and an oxidation of membrane thiol groups.

MATERIALS AND METHODS

Rat liver mitochondria were prepared by the method of Schneider (25). Changes in Ca²⁺concentration in the suspending medium were followed using a Ca²⁺-selective electrode (Radiometer, F 2112 Calcium Selectrode) calibrated with internal standards added to the reaction medium. Pyridine nucleotide fluorescence (excitation at 366 nm, emission at 450 nm) was monitored in an Aminco-Bowman Spectrophotofluorometer. Small structural changes of mitochondria, as measured by absorbance changes (19), were followed at 700 nm in the split-beam mode in an Aminco DW-2 dual wavelength spectrophotometer.

RESULTS AND DISCUSSION

Fig.1 shows the effect of bongkrekate and carboxyatractylate on the rate of mitochondrial Ca^{2+} efflux induced by the oxidized state of NAD(P). The upper traces show the NAD(P)H fluorescence in the presence of exogenous β -hydroxybutyrate or oxaloacetate included in the medium after Ca^{2+} uptake was

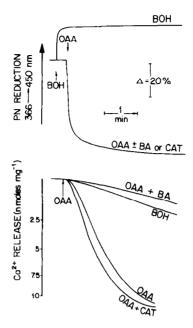


Figure 1. Effect of bongkrekate and carboxyatractylate on Ca $^{2+}$ $\overline{\rm eff1ux}$ induced by oxaloacetate. The system (2.0 ml, 35°C), contained: 125 mM sucrose, 60 mM KC1, 20 mM K*-acetate, 3.0 mM Hepes buffer, pH 7.1, 1.0 mM phosphate, 4.0 μM rotenone, 0.1 nmol antimycin A mg $^{-1}$, 1.0 μg oligomycin mg $^{-1}$, 0.5 mM Mg $^{2+}$ and 4.0 mg mitochondrial protein. TMPD (12 μM) and ascorbate (1.0 mM) were used as respiratory substrates. Ruthenium red (0.7 μM) was used to interrupt Ca $^{2+}$ uptake. Bongkrekate (7.0 μM) and carboxyatractylate (20 μM) when present were added prior to the addition of mitochondria. β -hydroxybutyrate (3.0 mM) and oxaloacetate (0.5 mM) were added at the points indicated by the arrows.

interrupted by ruthenium red (not shown). The addition of β -hydroxybutyrate increased the reduction of pyridine nucleotides while oxaloacetate caused a large oxidation of these nucleotides notwithstanding the presence of bongkrekate or carboxyatractylate. The lower traces show that in the presence of β -hydroxybutyrate the ruthenium red-insensitive Ca^{2+} efflux occurred at a rate of 1 nmol $\min^{-1} mg^{-1}$. In the presence of oxaloacetate, however, the rate of Ca^{2+} efflux was increased to about 7 nmol Ca^{2+} $\min^{-1} mg^{-1}$. In spite of the lack of effect of bongkrekate and carboxyatractylate on the oxidation of NAD(P)H by oxaloacetate, it was observed that the former completely prevents, and the latter slightly stimulates, the ruthenium-red insensitive Ca^{2+} efflux. It has been proposed that the effect of these agents

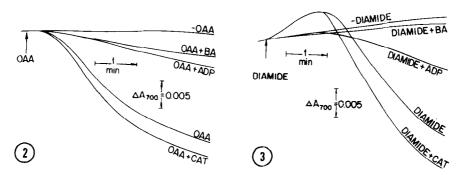


Figure 2. Changes in turbidity of the mitochondrial suspension induced by oxaloacetate. Mitochondria (5.0 mg protein) were preincubated in a system (2.5 ml, 30°C) containing: 250 mM sucrose, 3.0 mM Hepes buffer pH 7.1, 1.0 mM phosphate, 0.4 mM EGTA, 4.0 μM rotenone, 0.1 nmol antimycin A mg $^{-1}$, 1.0 μg oligomycin mg $^{-1}$. Oxaloacetate (1.0 mM) was added 5 minutes after the mitochondrial preincubation. Bongkrekate (10 μM), carboxyatractylate (20 μM) and ADP (200 μM) when present were added prior to the addition of mitochondria. Figure 3. Changes in turbidity of the mitochondrial suspension induced by diamide. Experimental conditions were exactly as in figure 2. Diamide (0.3 mM) was added at the point indicated by the arrow.

on Ca²⁺ efflux from mitochondria is related to the inhibition by bongkrekate and stimulation by carboxyatractylate of the Ca2+ -induced configurational transition of mitochondria (16). Related to this possibility, fig.2 shows that addition of oxaloacetate induced a decrease in absorbance ($\Delta A < 0.05$) of the mitochondrial suspension compatible with a small change in mitochondrial configuration (19). It can be seen that bongkrekate, which inhibits Ca²⁺ efflux, also inhibited the decrease in absorbance induced by oxaloacetate. ADP, another inhibitor of Ca2+ efflux (23) showed a similar effect. On the other hand, carboxyatractylate, which stimulates Ca2+ efflux, increased the change in absorbance induced by oxaloacetate. To avoid the interference of the configurational changes of mitochondria induced by Ca²⁺ accumulation (16-18) or energization (26), in these experiments the mitochondria were preincubated under de-energized conditions in the presence of EGTA.

The experiments in fig. 3 show the effect of diamide, an -SH oxidant, on the turbidity of the mitochondrial suspension.

The comparison between the effects of oxaloacetate and diamide was aimed at examining the possible participation of thiol groups on the mechanism of NADP +-induced Ca 2+ efflux from mitochondria since the oxidation of pyridine nucleotides may lead to an oxidation of critical membrane -SH groups by decreasing the availability of reduced glutathione in the matrix as shown by Bellomo et al (27). Like oxaloacetate, diamide induced a decrease in absorbance of the mitochondrial suspension at 700 nm. Most striking was the similarity between the effects of bongkrekate, ADP and carboxyatractylate on the decrease of absorbance induced by both diamide (fig.3) and oxaloacetate (fig.2). This similarity strongly suggests that the changes in mitochondrial configuration and the Ca2+ release induced by NAD(P) + and diamide have a common denominator. The fact that diamide is an oxidizing agent of membrane-bound thiol groups (24) indicates that the oxidation of NADPH may also cause an oxidation of critical membrane -SH groups that is linked to the oxidation of pyridine nucleotides as proposed by Harris et al (22) and Jung and Brierley (15). These results are in line with cumulative evidence that agents which react with thiol groups induce Ca²⁺ efflux from mitochondria, such as diamide (24), mercurials (22), quinones (27), and t-butyl hydroperoxides (4), and provides an indication for the participation of a membrane component in the mechanism of NAD(P) +-induced Ca2+ efflux from mitochondria. The alterations in mitochondrial structure observed under these experimental conditions are of the same order of magnitude of those induced by the binding of bongkrekate or atractylate to the AdN-translocase in heart mitochondria (19) and certainly do not reflect gross membrane damage since our recent work (29) shows that at low Ca2+ concentrations the NADP+-induced Ca2+ efflux from mitochondria is completely reversible.

ACKNOWLEDGEMENTS

We thank Dr. Albert L. Lehninger for the use of his dual-wavelenght spectrophotometer. Bongkrekic acid was a gift from Dr. Pierre Vignais to Dr. Lehninger. This work was supported in part by grants from FAPESP (Proc. 83/1760-1) and CNPg (Proc. 11233/82).

REFERENCES

- Lehninger, A.L., Vercesi, A.E. and Bababunmi, E.A. (1978) Proc. Natl. Acad. Sci. USA 75, 1690-1694.
- Coelho, J.L.C. and Vercesi, A.E. (1980) Arch. Biochem. 2. Biophys. 204, 141-147.
- Dawson, A.P., Selwyn, M.J. and Fulton, D.V. (1979) Nature, 3. 277, 484-486.
- Lotscher, H.R., Winterhalter, K.H., Carafoli, E. and Richter, C. (1979) Proc. Natl. Acad. Sci. USA 76, 4340-4344.
- Beatrice, M.C., Palmer, J.W., Pfeifer, D.R. (1980) J.Biol. Chem. 255, 8663-8671.
- Nicholls, D.G. and Brand, M.D. (1980) Biochem.J. 188, 113-118.
- Panfili, E., Sottocasa, G.L., Sandri, G. and Liut, G. (1980) Eur. J. Biochem. 105, 205-210.
- Prpic, V. and Bygrave, F.L.(1980)J.Biol.Chem. 255,6193-6199.
- Moore, G.A., Jewell, S.A., Bellomo, G. and Orrenius, S. (1983) FEBS Lett. 153, 289-292. 9.
- 10. Bardsley, M.E. and Brand, M.D. (1982) Biochem.J. 202, 197-201.
- Krell, H., Baur, H., Pfaff, E. (1979) Eur.J.Biochem. 101, 11. 349-364.
- Bellomo, G., Jewell, S.A., Thor, H. and Orrenius, S. (1982) 12. Proc. Natl. Acad. Sci. USA 79, 6842-6846.
- 13. Sies, H., Graf, P. and Estrela, J.M. (1981) Proc. Natl. Acad. Sci. USA 78, 3358-3362.
- 14. Hofstetter, W., Muhlebach, T., Lotscher, H.R., Winterhalter, K.H. and Richter, C. (1981) Eur.J.Biochem. 117, 361-367.
- 15. Jung, D.W. and Brierley, G.P. (1982) Biochem. Biophys. Res. Communs. 106, 1372-1377.
- Hunter, D.R. and Haworth, R.A. (1979) Arch. Biochem. Biophys. 16. 195, 453-459.
- 17. Haworth, R.A. and Hunter, D.R. (1979) Arch.Biochem.Biophys. 195, 460-467.
- Hunter, D.R. and Haworth, R.A. (1979) Arch. Biochem. Biophys. 18. 195, 468-477.
- Scherer, B. and Klingenberg, M. (1974) Biochemistry, 13, 19. 161-170.
- Out, T.A., Kemp, Jr. A. and Souveriyn, J.H.M. (1971) Biochim. 20. Biophys. Acta 245, 299-304.
- Peng, C.F., Straub, K.D., Kane, J.J., Murphy, M.L. and 21. Wadkins, C.L. (1977) Biochim. Biophys. Acta, 462, 403-413.
- Harris, E.J., Al-Shaikhaly, M. and Baum, H. (1979) Biochem. J. 182, 455-464. 22.
- Zoccarato, F., Rugolo, M., Siliprandi, D. and Siliprandi, N. 23. (1981) Eur. J. Biochem. 114, 195-199.
- Rugolo, M., Siliprandi, D., Siliprandi, N. and Toninello, A. 24. (1981) Biochem.J. 200, 481-486. Schneider, W.C. (1948) J.Biol.Chem. 176, 259-266.
- 25.
- Hackenbrock, C.R. (1966) J.Cell.Biol. 30, 269-297. 26.
- Bellomo, G., Jewell, S. and Orrenius, S. (1982) J.Biol. 27. Chem. 257, 11558-11562.
- Harris, E.H. and Baum, H. (1980) Biochem.J. 186, 725-732. 28.
- Vercesi, A.E. and Lehninger, A.L. (1983) J.Biol.Chem., in press. 29.